

Mycosphere 7 (6): 844–852 (2016) www.mycosphere.org Copyright © 2016

# **Article**

ISSN 2077 7019 Mycosphere Online Edition

Doi 10.5943/mycosphere/7/6/15

# Extracellular synthesis of silver nanoparticles using *Aspergillus* versicolor and evaluation of their activity on plant pathogenic fungi

Elgorban AM<sup>1,2</sup>, Aref SM<sup>2</sup>, Seham SM<sup>2</sup>, Elhindi KM<sup>3</sup>, Bahkali AH<sup>1</sup>, Sayed SR <sup>4</sup>, Manal MA<sup>1</sup>

Elgorban AM, Aref SM, Seham SM, Elhindi KM, Bahkali AH, Sayed SR, Manal MA 2016 – Extracellular synthesis of silver nanoparticles using *Aspergillus versicolor* and evaluation of their activity on plant pathogenic fungi. Mycosphere 7(6), 844–852, Doi 10.5943/mycosphere/7/6/15

## **Abstract**

In the current study, silver nanoparticles (Ag NPs) were synthesized using *Aspergillus versicolor*. The synthesized Ag NPs were characterized by UV–Vis spectrum, TEM, XRD, SEM, and EDS, which revealed that the synthesized NPs had a face-centred cubic similarity. The rapid synthesis of Ag NPs in fungal filtrate showed bright sunlight. The maximum absorbance of Ag NPs was observed at 620 nm which is a sign of Ag NPs. The TEM analysis revealed the spherical shape with the size ranged between 5 to30 nm and EDS showed the presence of Ag at 3kev. The antifungal activity of biogenic Ag NPs was evaluated against white mold (*Sclerotinia sclerotiorum*) and grey mould (*Botrytis cinerea*) in strawberry (*Fragaria x ananassa*). The results showed that synthesized silver nanoparticles exhibit significant antifungal potential on both pathogens in a concentration-dependent manner. The maximum reduction in both fungi was observed at 150 ppm of Ag NPs.

**Key words** – Aspergillus versicolor – biomass – grey mould – Sclerotinia sclerotiorum – silver nanoparticles

## Introduction

Agricultural output is decreased worldwide year after year because of plant diseases. Millions have been expended in efforts to protect these plant diseases. Several techniques have been applied to control phytopathogenic fungi. The use of fungicides is the main widespread method to control the diseases (Zhang et al. 2007). In current years, environmental threats caused by excessive use of fungicides have been discussed extensively. Therefore, the scientist and researchers in the agricultural area are searching for alternatives in synthetic fungicides. As an alternative to agrochemicals, utilization of silver nanoparticles as antifungal agents has become more prevalent as advanced technologies make their manufacture more economical (Kim et al. 2012, Elgorban et al. 2016). One of the possible applications of Ag is in the control of plant pathogenic fungi. As silver has been shown various modes of inhibitory effect against several microorganisms (Jin-Hee et al. 2010). Thus, it could be utilized with comparative safety for management of several plant diseases as compared to agrochemicals (Kabir et al. 2011a, b). In earlier studies the eco-friendly properties

<sup>&</sup>lt;sup>1</sup>Botany and Microbiology Department, College of Science, King Saud University, Saudi Arabia.

<sup>&</sup>lt;sup>2</sup>Plant Pathology Research Institute, Agricultural Research Center, Giza, Egypt.

<sup>&</sup>lt;sup>3</sup>Plant Production Department, College of Food and Agricultural Sciences, King Saud University, Saudi Arabia.

<sup>&</sup>lt;sup>4</sup>Department of Zoology, College of Science, King Saud University, Saudi Arabia

and intense effects of Ag+ have been shown. Major requirements for the possible utilization of Ag in management of plant disease involve the need for further studies on antimicrobial activities of several silver compounds against phytopathogenic fungi and to improve the efficiency of disease management (Kim et al. 2012). The green synthesis of silver nanoparticles is in the limelight in new nanotechnology (Gade et al. 2009, Elgorban et al. 2016, Shaban et al. 2015). The culture of fungi at large scale is simple, whereas, the bacterial system consists utilization of progressing instruments to achieve pure filtrate from colloidal broth. In the fungal system, it secrets extracellular enzymes, which have extra features in the downstream treatment and handling of biomass (Gade et al. 2008). In view of fungi properties, it may be exceedingly used for the fast and eco-friendly green synthesis of mineral NPs.

Green synthesis of Ag NPs by using several fungi such as the bioagent *Trichoderma* (Singh et al. 2011, Vahabi et al. 2011), *Aspergillus* (Patil et al. 2011, Li et al. 2013), and *Fusarium* (Anitha & Palanivelu, 2011, Soni & Prakash 2011) has been reported. Newly the antifungal activity of silver nanoparticles has been studied by several researchers (Panáček et al. 2009, Qian et al. 2013, Roy et al. 2013, Xu et al. 2013, Elgorban et al. 2016). Conversely, few studies are available on the antifungal activity of Ag nanoparticles against plant pathogenic fungi (Velmurugan et al. 2009, Kasprowicz et al. 2010, Lee et al. 2013). Strawberry (*Fragaria ananassa*) is one of the most extensively grown small fruit crops in the world. *Botrytis cinerea* and *Sclerotinia sclerotiorum* are responsible for severe economic losses (Williamson et al. 2007, Mertely & Peres, 2009), in some places, the yield losses reaching to 90% (Ellis & Grove 1982). Strawberry fruits have a very short postharvest life, gray mold caused by *B. cinerea* and white mold caused by *S. sclerotiorum* infection may occur and cause fruit decay.

Therefore, this study has been focused on synthesis of Ag NPs using *Aspergillus versicolor* and also the inhibitory effect of these silver nanoparticles against white mold (*S. sclerotiorum*) and grey mold (*B. cinerea*) in strawberry (*Fragaria x ananassa*) in vitro.

## **Materials & Methods**

## **Isolation and identification of microorganisms**

## Aspergillus versicolor

Aspergillus versicolor was isolated from soil obtained from the greenhouses, Riyadh, Saudi Arabia and was identified according to Benndorf et al. (2008).

## The pathogenic fungi

Sclerotinia sclerotiorum and Botrytis cinerea were isolated from strawberry (Fragaria x ananassa). The fungi were cultured in potato dextrose agar (PDA) medium. On the basis of cultural properties and microscopic observations, S. sclerotiorum and B. cinerea were identified as described (Ai-Rong et al. 2008; Chilvers & du Toit 2006).

## **Synthesis of silver nanoparticles**

To prepare silver nanoparticles, *A. versicolor* was grown aerobically in a liquid medium containing: glucose, 15.0 g; KH<sub>2</sub>PO<sub>4</sub>, 7.0 g; K<sub>2</sub>HPO<sub>4</sub>, 2.0 g; (NH<sub>2</sub>)SO<sub>4</sub>, 1.0 g; MgSO<sub>4</sub>·7H<sub>2</sub>O, 0.1 g; and yeast extract, 1.0 g. The Erlenmeyer flasks were inoculated with a spore suspension of *A. versicolor* and were incubated on an incubator shaker at 25°C (120 rpm). The biomass was collected after 7 days by filtration through the filter paper (Whatman No. 1), washed four times with sterilized distilled water to eliminate any medium component. The obtained biomass (20g, fresh weight) was mixed with 150 mL of distilled water for 48 h at 25°C (120 rpm). The cell filtrate was detached from flasks by passing it through Whatman No. 1. For NPs synthesis, silver nitrate, 1 mM final concentration (50 or 100 mL) was mixed with 50 or 100 mL of the cell filtrate in a flask and incubated at 35 °C (120 rpm) in the dark for 5 days. Filtrates without the silver ion were served as control. All the tests were done in triplicates.

## **Characterization of Ag NPs**

The qualitative experiment for affirmation of Ag NPs was carried out with UV-vis spectroscopy. 1 mL of specimen aliquot from this bio transformed product was drawn after 24 post incubation with AgNO3 solution, and absorbance was noted by using Biochrom Anthos 2020 spectrophotometer wavelength range: 400 - 750 nm, with 4 standard filters (405, 450, 492, 620 nm), including onboard software to examine the change in light absorption of the solution with rise in color intensity. The small amount of the dried precipitate was spared on carbon tap with stub. The specimen was subjected to SEM (JEOL, JSM-6380 LA, Japan). The intermediate shapes and size of the Ag NPs were measured by transmission electron microscopy (TEM). A drop of NPs suspension was put on a carbon-coated copper grid, dried in oven at 65 h for 24h. Micrographs were obtained in transmission electron microscope (JEOL Ltd., Akishima-shi, Japan) with 80- to 120-kV accelerating voltage at 0.23-nm resolution.

# Antifungal activity of silver nanoparticles on Sclerotinia sclerotiorum and Botrytis cinerea

In vitro test was conducted on PDA treated with various concentrations (25, 50, 100, and 150 mg/L) of silver nanoparticles. Five mL of silver nanoparticles having various concentrations was poured into PDA prior to plating in a Petri plate. PDA containing Ag NPs was incubated at 20±2°C. After 2 days of incubation, agar disc (5 mm) containing fungi were inoculated simultaneously at the center of each Petri plates containing Ag NPs, followed by incubation at 20±2°C for 5 days. The following formula was utilized for the calculation of the inhibition rate (%).

Inhibition rate 
$$\% = \frac{M-m}{M} \times 100$$

Where M is the radial growth of fungal mycelia on the control plate and m is the radial growth of fungal mycelia on the plate with silver nanoparticles.

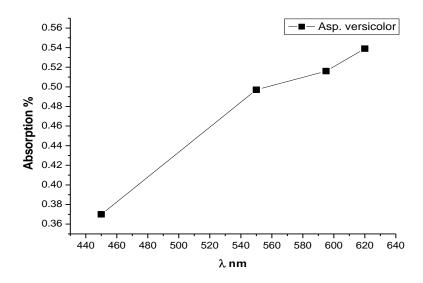
#### Results and discussion

# UV-vis spectrum of Ag NPs

Aspergillus versicolor was grown on culture liquid medium. The biomass was collected to obtain mycelia free cell filtrate. It was observed that the color was changed after the addition of silver nitrate to free cell filtrate by visual notice in the culture. The color of cell filtrate containing silver nitrate changed from colorless to yellow after 10 min and finally to brown. This change in color refers to the formation of Ag NPs in the solution, while free cell silver nitrate did not display any visual color change (Fig. 1). The increase in the color density also depends on the size of the synthesized silver nanoparticles. Smitha et al. (2009) confirmed that the color turn during metal NPs synthesis is due to the collective vibration of free conduction electrons stimulated by an interactive electromagnetic field that called surface plasmon peak. The UV absorption spectral studies were done to confirm the formation of Ag NPs using *A. versicolor* (Fig. 2). The absorption peak in the UV spectrum equal to the surface plasmon resonance, the maximum synthesis of Ag+was shown at 620 nm where maximum peak was found (Bhimba et al. 2014, Ninganagouda et al. 2014).



Fig. 1 – Color of A. versicolor free cell extract without AgNO<sub>3</sub> (A) and with AgNO<sub>3</sub> (B)



**Fig. 2** – UV-visible spectra recorded for the reaction of *A. versicolor* cell filtrate with AgNO3 solution.

## Transmission electron microscopy (TEM)

Transmission electron microscopy method was utilized to conceive the morphology of Ag NPs synthesized. As shown in Fig. 3, revealed individual instead of a number of assembled spherical Ag NPs. The NPs were not in direct contact even in the assembles and were enclosed by a thin layer of organic materials (Jagtap & Bapat 2013). The sizes of the particle histogram of Ag NPs are shown in Fig. 4. The results revealed that there was difference in both the average size and the particle size. The size of the particles was found in the ranges between 5 to 30 nm. The particle distribution presented in Fig 4, displays that the maximum percentage noticed of the particle size are in the range of 5 to 10 nm. Shekhar et al. (2014) revealed that the shape and size of the green synthesized Ag NPs depend on the pH and temperature of the medium as well as the microorganisms. The difference size of Ag NPs which produce by Aspergillus species have been reported as *Aspergillus terreus* (10–19 nm) Eepsita at al. 2014), *Aspergillus niger* (5– 35 nm) (Lamabam and Joshi, 2015), *Aspergillus* foetidus (20-40 nm) (Roy et al. 2013), and *Aspergillus parasiticus* (less than 50 nm) (Moazeni et al. 2014).

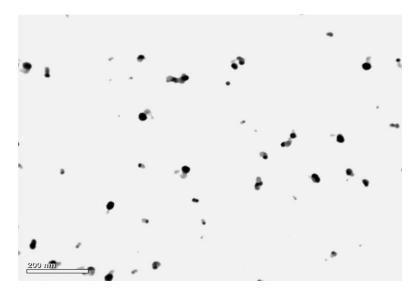
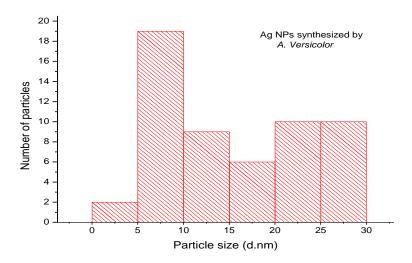


Fig. 3 – TEM micrographs of *silver* nanoparticles synthesized *A. versicolor* 



**Fig. 4** – A particle size distribution histogram of synthesized Ag NPs determined from Transmission Electron Microscopy (TEM) image.

## Scanning electron microscope (SEM) and energy dispersive X-ray analysis (EDS)

SEM images display irregular and spherical shaped silver nanoparticles with diverse sizes (Fig. 5). It was observed that there were presences of several silver nanoparticles clusters, which might be attributed to the aggregation of nanoparticles formed throughout specimen preparation (Puchalski et al. 2007, Selvi & Sivakumar 2012). Fig. 6 showed the EDS analysis of the NPs. The elemental structure of powdered specimen was determined using SEM equipped with an EDS detector. The energy dispersive X-ray analysis displayed the strong signal at about 3 keV of the Ag regions due to surface plasma resonance. Also, the ghostly signals for oxygen and carbon were noticed which indicated that the extracellular organic moieties from culture filtrate were adsorbed on the surface of the NPs (Bhainsa & D'Souza, 2006, Lamabam & Joshi 2015).

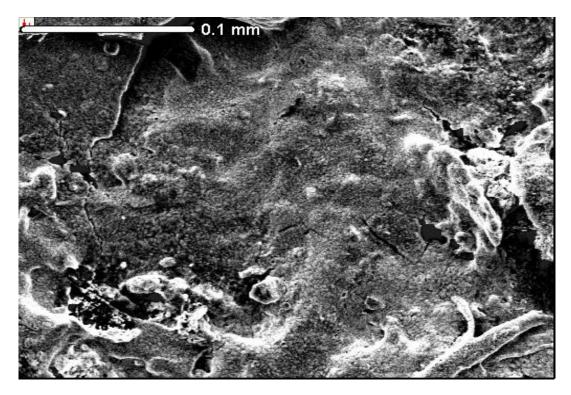


Fig. 5 – Scanning electron microscopy of Ag NPs formed by A. versicolor

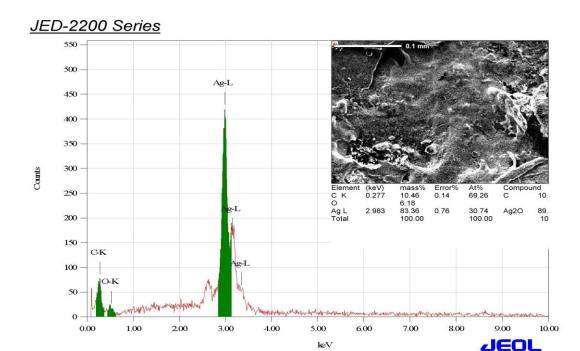


Fig. 6 – EDS spectrum recorded of silver nanoparticles formed by A. versicolor

# Antifungal activity of silver nanoparticles on Sclerotinia sclerotiorum and Botrytis cinerea

The inhibition effect of silver nanoparticles at various concentrations was analyzed in PDA (Table 1). The maximum inhibition of fungal growth was observed at 150 ppm. The highest value of inhibition was noticed against B. cinerea at 150 ppm that produced 80.38% inhibition in fugal growth with 72.4, 382.8 and 2.3±3.9 ED50, ED95, and Slope±SE, respectively. Also, the highest level of growth inhibition of S. sclerotiorum was observed at 150 ppm giving 74.39% reduction in fungal growth with 96.5 (ED50), 591.1 (ED95), and 2.1±4.1 (Slope±SE). The present study demonstrated that silver nanoparticles were found to be very effective against both plant pathogenic fungi. The results suggested that silver nanoparticles have the capacity to inhibit the growth of S. sclerotiorum and B. cinerea. The effect was observed in a concentration dependent manner in both fungi. The both fungi showed maximum inhibition at 150 ppm Ag NPs concentration. The inhibition was increased as the concentration of silver nanoparticles increases. This high antifungal activity of silver nanoparticles is probably related to the high intensity at which Ag NPs solution were capable of state and agglutinate to fungal hyphe and to deactivate plant pathogenic fungi. There are many mechanisms of inhibitory effect of Ag+ on microorganisms, such as DNA loses its ability to replicate (Feng et al. 2000), resulting in inactivated expression of ribosomal subunit proteins, instead of some other enzymes and cellular proteins necessary to the adenine triphosphate production (Yamanaka et al. 2005). In addition, it has been supposed that silver ion primarily affects the membrane-bound enzymes function, for instance those in the respiratory chain (Bragg & Rainnie 1974, McDonnell & Russell 1999).

**Table 1** Antifungal activity of extracellular silver nanoparticles against plant pathogenic fungi (Inhibition %)

Organisms	Concentration (mg/L)					ED50	ED95	Slpoe±SE
	0	25	50	100	150	2200	2270	~- <b>r</b> • •====
S. sclerotiorum	82.00	15.24	22.87	42.07	74.39	96.5	591.1	2.1±4.1
B. cinerea	79.00	16.77	33.54	58.23	80.38	72.4	382.8	$2.3\pm3.9$

In outline, silver nanoparticles have great potential antifungal effects of both fungi tested, perhaps throughout the demolition of membrane solidity; consequently, it was concluded that silver nanoparticles have significant antifungal activity.

## **Disclosure statement**

No potential conflict of interest was reported by the authors.

# Acknowledgments

The authors extend their sincere appreciations to the Deanship of Scientific Research at King Saud University for its funding this Prolific Research Group (PRG-1437-34).

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